

EDVO-Kit: AP05

Photosynthesis

See Page 3 for storage instructions.

EXPERIMENT OBJECTIVE:

In this experiment, students will learn how to measure the rate of photosynthesis indirectly by using the floating leaf disk method. They will also investigate several factors that might affect the photosynthesis process.

1.800.EDVOTEK

www.edvotek.com

info@edvotek.com

Table of Contents

	Page
Experiment Components	3
Experiment Requirements	3
Background Information	4
Experiment Procedures	6
Experiment Overview	6
Investigation I: Observation of environmental factors that affect the rate of photosynthesis	7
Investigation II: Observation of plant respiration	10
Data Collection and Analysis	12
Study Questions	14
Instructor's Guidelines	
Notes to the Instructor	15
Pre-Lab Preparations	16
Experiment Results and Analysis	17
Study Questions and Answers	19
Material Safety Data Sheets	20

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All components are intended for educational research only. They are not to be used for diagnostic or drug purposes, nor administered to or consumed by humans or animals.

THIS EXPERIMENT DOES NOT CONTAIN HUMAN DNA. None of the experiment components are derived from human sources.

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Experiment Components

- Sodium Bicarbonate (baking soda)
- Liquid soap
- Plastic syringes
- Transfer pipets
- Plastic cups

Store the entire experiment at room temperature.

This experiment is designed for 10 lab groups.

Requirements

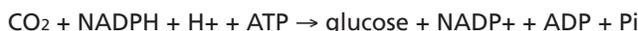
- Leaves (i.e. spinach, ivy, pokeweed)
- Timer
- Light source (60 watt light bulb recommended)
- Hand-held hole punch
- Beakers

Background Information

Photosynthesis is the process by which plant cells use light energy for the biosynthesis of cellular components. Photosynthetic organisms form the basis of the food chain. These life forms include higher plants, algae, dinoflagellates, euglenoids, diatoms and certain bacteria. Photosynthesis consists of two biochemical phases. The general equation for the first phase is:



The first phase is light dependent. NADP⁺ and NADPH are oxidized and reduced forms, respectively, of nicotinamide adenine dinucleotide phosphate. The reduced form is an essential cofactor in the biosynthesis of many types of molecules such as carbohydrates. Chemical energy of ATP is required for many biochemical reactions and for maintenance of cellular integrity and function. ATP is generated from ADP and inorganic phosphate (P_i). The reaction also generates protons (H⁺) and molecular oxygen from water. The second phase of photosynthesis can be generally written as:



The second phase is not light dependent. The reaction fixes atmospheric carbon dioxide into organic linkage (glucose). Each phase consists of many separate chemical steps. First phase steps are called light reactions and second phase steps are called dark reactions.

Light reactions in eukaryotic cells occur in organelles called chloroplasts. Chloroplasts contain DNA and are self-replicating. These organelles consist of an outer membrane and a folded inner membrane. Stacked, disk-like structures called thylakoids form part of the inner membrane and it is here that light dependent photosynthetic systems are found. The primary photosynthetic pigments are green chlorophylls. Chloroplasts contain chlorophyll 'a' and 'b', magnesium-porphyrin complexes, and are specifically bound to proteins that reside on and within the inner membrane. Pure chlorophyll 'a' maximally absorbs light at wavelengths of around 420 and 660 nm. Chlorophyll 'b' absorbs primarily at approximately 480 and 640 nm.

The absorption spectrum of chlorophylls can be shifted depending on with which type of protein they are associated. Other pigments found in chloroplasts include β-carotene and xanthophylls. These pigments have an accessory light harvesting function and absorb at wavelengths in between the maxima of the chlorophylls. They all capture light energy and transfer it to the chlorophyll a at the reaction center. All these pigments are sensitive to light and oxygen in the purified state and eventually breakdown. Extremely pure preparations required for chemical and biological studies are stored under vacuum, in the dark, at -20° C.

Photosynthetic System

There are two photosynthetic systems in chloroplasts, termed Photosystem i and ii. These physically distinct systems contain different proteins and ratios of chlorophylls and accessory pigments. Photosystem i is not responsible for oxygen evolution and is activated by longer wavelengths of light. Photosystem ii is activated by shorter wavelengths of light and is required for oxygen and ATP production. Both systems contribute high energy electrons for the reduction of NADP⁺. Both photosystems are required for maximal photosynthetic activity. When light is absorbed by the chlorophyll-protein complex of photosystem I, chlorophyll becomes excited and enters a higher energy state. During return



Background Information

from high energy to ground state, an electron is boosted to a higher energy level and is sequentially transferred, via several membrane associated protein transport molecules, to the final electron acceptor NADP⁺. The NADP⁺ is reduced to NADPH through the action of a reductase enzyme. Since this is not normally a spontaneous (energetically favorable) reaction, input of light energy is required to convert NADP⁺ to NADPH.

Electron transport proteins containing iron and sulfur are called ferredoxins. Other transporters are called cytochromes. A high energy electron generated by light absorption in photosystem II is donated, via a specific sequence of transporters, to the electron deficient photosystem I. Photosystem II then receives an electron through a series of transport proteins from H₂O. Water is oxidized to molecular oxygen during this process. Water is the electron donor in photosynthesis. Conversion of water, a very stable molecule, to oxygen is energetically unfavorable and would not occur to any significant extent without input of light energy at photosystem II. Production of ATP occurs along the sequence of electron transfer steps. ATP is a high energy compound and requires energy for its synthesis. This energy is siphoned from the high energy electrons through a complex series of events involving membrane proteins and formation of pH gradients across chloroplast membranes to drive ATP synthesis.

The dark reactions of photosynthesis occur simultaneously with the light reactions in plant cells. The dark reactions are a set of seven enzyme catalyzed metabolic steps that synthesize glucose. The key metabolic step in plant glucose synthesis is catalyzed by the abundant enzyme ribulose diphosphate carboxylase. The majority of these reactions take place outside the chloroplast in the cytoplasm. Most of the glucose is polymerized into starch and cellulose. The reaction involves the fixation of carbon dioxide.

This experiment uses the floating disk leaf assay to explore the process of photosynthesis in plants. Leaf disks generally float due to the many intercellular spaces used for exchange of gases. When the air spaces are infiltrated with solution, the overall density of the leaf disk increases, causing the disks to sink. The infiltration solution includes a small amount of Sodium bicarbonate. Bicarbonate ion serves as the carbon source for photosynthesis. By providing the components needed for photosynthesis (light, CO₂, and H₂O), oxygen will be produced in the leaf. As photosynthesis takes place, oxygen is released into the interior of the leaf which changes the buoyancy and causes the disks to rise. Since cellular respiration takes place and also consumes oxygen, the rate that the disks rise is an indirect measurement of the net rate of photosynthesis.

Respiration, which uses the oxygen produced by photosynthesis, is also observed in the leaf disk assay. Some of the oxygen will be used in the leaf's respiration process. Leaf disks float, because the net result is that more oxygen is produced by photosynthesis than is used in respiration. In this investigation, the rate at which leaf "disks" rise will be used as an indirect measure of the net production of oxygen produced by photosynthesis.

Experiment Overview and General Instructions

EXPERIMENT OBJECTIVE:

In this experiment, students will learn how to measure the rate of photosynthesis indirectly using the floating leaf disk method. They will also investigate several factors that might affect the photosynthesis process.

LABORATORY SAFETY GUIDELINES

1. Wear gloves and goggles while working in the laboratory.
2. Exercise caution when working in the laboratory – you will be using equipment that can be dangerous if used incorrectly.
3. DO NOT MOUTH PIPET REAGENTS - USE PIPET PUMPS.
4. Always wash hands thoroughly with soap and water after working in the laboratory.
5. If you are unsure of something, ASK YOUR INSTRUCTOR!



LABORATORY NOTEBOOKS:

Scientists document everything that happens during an experiment, including experimental conditions, thoughts and observations while conducting the experiment, and, of course, any data collected. Today, you'll be documenting your experiment in a laboratory notebook or on a separate worksheet.

Before starting the Experiment:

- Carefully read the introduction and the protocol. Use this information to form a hypothesis for this experiment.
- Predict the results of your experiment.

During the Experiment:

- Record your observations.

After the Experiment:

- Interpret the results – does your data support or contradict your hypothesis?
- If you repeated this experiment, what would you change? Revise your hypothesis to reflect this change.

Investigation I: Observation of Cellular Photosynthesis**Notes:**

- For Module I, you will receive 2 plastic cups – (1) cup containing water / liquid soap solution, and (1) cup containing the bicarbonate / liquid soap solution.
- Throughout this module, you will be preparing material for the “Light - Control soln” and “15 cm Light - CO₂ soln” simultaneously. Follow the steps below for preparation of materials.

Procedure

1. Label the plastic cup containing water / liquid soap solution provided by your lab instructor as “Light - Control soln.” This is your control cup.
2. Label the plastic cup containing the bicarbonate / liquid soap solution provided by your lab instructor as “15 cm Light - CO₂ soln.” This is your experimental cup.
3. Prepare 10 uniform leaf disks for each trial using the hole punch. Try not to include the major veins in the leaf disks (Fig. 1).
4. Remove the plunger of the syringe and carefully transfer leaf disks in the barrel. Shake or tap the barrel on the lab bench to collect the disks to the bottom (near the opening) of the barrel.
5. Replace the plunger back into the barrel. Push the plunger until only a small volume of air and leaf disks remain in the barrel (Fig. 2). Be careful not to damage the leaf disks.
6. Using the transfer pipet provided, put a small volume of sodium bicarbonate solution (4-5 cc) into the syringe. Gently shake or tap the syringe to suspend the leaf disks in the solution.
7. While placing a finger over the syringe opening tightly, draw back slowly on the plunger to create a vacuum and hold it for 10 seconds (Fig. 3).
8. While holding the vacuum, swirl the leaf disks to suspend them in solution. Turn the syringe upright and slowly let the plunger spring back to release the vacuum.
9. If the disks don't sink, repeat steps 7-8. You may have to repeat this procedure two to three times in order to get all the disks to sink to the bottom (near the opening) of the syringe barrel (Fig. 4).



Fig. 1 - Making leaf disks.



Fig. 2 - Placing leaf disks into the syringe.



Fig. 3 - Creating vacuum in the plunger.



Fig. 4 - Sinking leaf disks.

Investigation I: Observation of Cellular Photosynthesis, continued



Fig. 5 - Sinking leaf disks in bicarbonate solution.



Fig. 6 - Placing cup under light source.



Fig. 7 - Disks floating in bicarbonate solution.

- Once all the disks have sunk, remove plunger from the barrel. Swirl and quickly pour the disks and the solution into the cup labeled "15 cm Light - CO₂ soln" (Fig. 5).

If the disks stick to the side of the syringe, add a small amount of Sodium Bicarbonate Solution into the syringe. Slowly swirl the syringe to dislodge the disks and pour it into the plastic cup (or beaker).

- Repeat steps 3 – 9 for the cup labeled "Light – Control soln." Remember to replace the bicarbonate solution with the diluted soap solution in the plunger.
- Place both cups under light located about 15 cm away and begin timing (Fig. 6).
- Use Table 1 to record the number of disks that are floating in the "Light – Control soln" cup at the end of each minute (Fig. 7).
- Use Table 2 to record the number of disks that are floating in the "15 cm Light - CO₂ soln" at the end of each minute.
- Continue to record the number of floating disks at the end of each minute until all of the disks are floating or you have reached 30 minutes.

Investigation I: Observation of Cellular Photosynthesis, continued

Table 1: Light – Control Solution	
Minutes	# of leaf disks floating
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
11	
12	
13	
14	
15	
16	
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18	
19	
20	
21	
22	
23	
24	
25	
26	
27	
28	
29	
30	

Table 2: 15 cm Light – Co₂ Solution	
Minutes	# of leaf disks floating
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
11	
12	
13	
14	
15	
16	
17	
18	
19	
20	
21	
22	
23	
24	
25	
26	
27	
28	
29	
30	

Investigation II - Observation of Plant Respiration**Question:**

What happens if photosynthesis is not occurring, but cellular respiration continues?

Notes:

- For Investigation II, you will receive 1 plastic cup containing the bicarbonate / liquid soap solution

Procedure

1. Label the plastic cup containing bicarbonate / liquid soap solution provided by your lab instructor as "30 cm Light - CO₂ soln."
2. Prepare 10 uniform leaf disks using the hole punch. Remember to avoid the major veins in the leaf when making leaf disks.
3. Remove the plunger of the syringe and carefully transfer your leaf disks in the barrel. Shake or tap the barrel on the lab bench to collect the disks to the bottom (near the opening) of the barrel.
4. Replace the plunger back into the barrel. Push the plunger until only a small volume of air and leaf disk remains in the barrel. Be careful not to damage the leaf disks.
5. Using the transfer pipet provided, put a small volume of sodium bicarbonate solution (4-5 cc) into the syringe. Gently shake or tap the syringe to suspend the leaf disks in the solution.
6. While placing a finger over the syringe opening tightly, draw back slowly on the plunger to create a vacuum and hold it for 10 seconds.
7. While holding the vacuum, swirl the leaf disks to suspend them in solution. Turn the syringe upright and slowly let the plunger spring back to release the vacuum.
8. If the disks don't sink, repeat steps 7-8. You may have to repeat this procedure two to three times in order to get all the disks to sink to the bottom.
9. Once all the disks have sunk, remove plunger from the barrel. Swirl and quickly pour the disks and the solution into the cup containing bicarbonate / liquid soap solution.

If the disks stick to the side of the syringe, add a small amount of Sodium Bicarbonate Solution into the syringe. Slowly swirl the syringe to dislodge the disks and pour it into the plastic cup (or beaker).
10. Place the cup under light located about 30 cm away and begin timing (Fig. 6).
11. Use Table 3 to record the number of disks that are floating at the end of each minute until all of the disks are floating or you have reached 30 minutes.
12. Once all the leaf disks have floated, remove the cup from the light source and place it in the dark. Suggestions include covering the beaker with an empty box or a piece of aluminum foil. Use Table 4 to record how many disks are still floating at the end of each minute over the next 15 minutes.

Investigation II - Observation of Plant Respiration

Table 3: 30 cm Light – Co₂ Solution	
Minutes	# of leaf disks floating
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
11	
12	
13	
14	
15	
16	
17	
18	
19	
20	
21	
22	
23	
24	
25	
26	
27	
28	
29	
30	

Table 4: Dark – Co₂ Solution	
Minutes	# of leaf disks floating
1	
2	
3	
4	
5	
6	
7	
8	
9	
10	
11	
12	
13	
14	
15	
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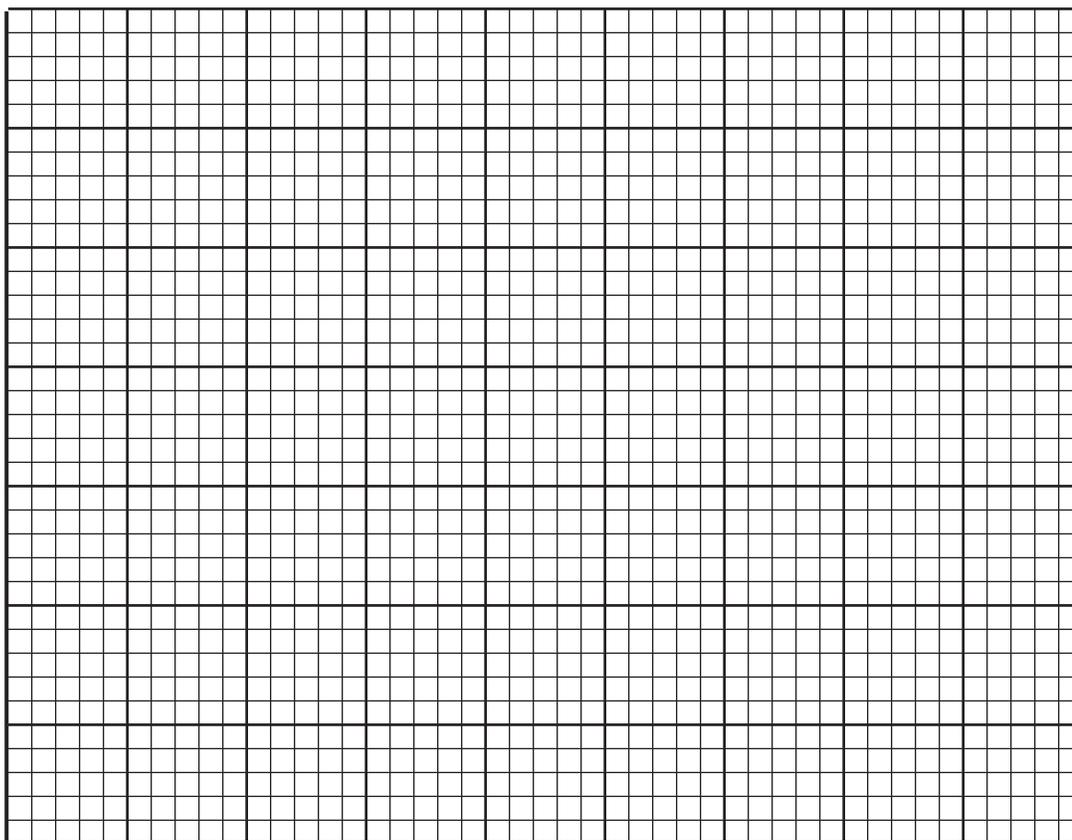
Data Collection and Analysis

Graph your results for **each** of the trials on the graph paper provided. Graph the results from your class average.

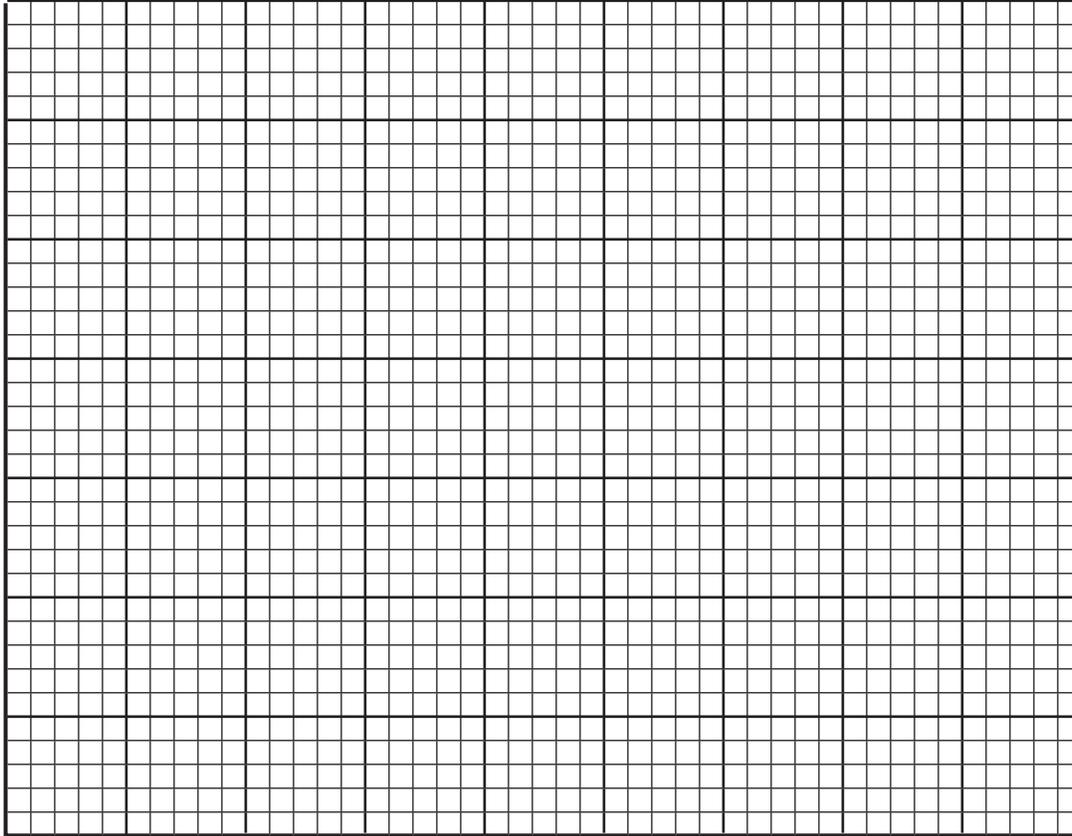
- Label the independent variable (horizontal x-axis).
- Label the dependent variable (vertical y-axis).
- Title the Graphs
- ET_{50light} is the point at which 50% of leaf disks are floating (the median). Find the ET_{50light} value of each trial (if applicable).
- ET_{50resp} is the time for 50% to sink after the leaf disks were transferred to the dark conditions. Determine the ET_{50resp} value.
- Because respiration occurs in both the light and dark, the rate of photosynthesis (ET_{ps}) is the sum of the rate in the light plus the respiration rate.

$$1/ET_{50ps} = 1/ET_{50light} + 1/ET_{50resp}$$

What is the rate of photosynthesis for your experiment?



Data Collection and Analysis



Study Questions

1. What is the function of the sodium bicarbonate in this experiment?
2. Explain the process that causes the leaf disks to rise.
3. Explain the process that causes the leaf disks to sink.
4. What is the purpose of using water/soap solution for one of the trials?
5. What is the effect of darkness on photosynthesis? Explain.
6. How does light intensity affect the rate of photosynthesis?



Instructor's Guide

Notes to the Instructor & Pre-Lab Preparations

Overview of Laboratory Investigations

The "hands-on" laboratory experience is a very important component of science courses. Laboratory experiment activities allow students to identify assumptions, use critical and logical thinking, and consider alternative explanations, as well as help apply themes and concepts to biological processes.

EDVOTEK experiments have been designed to provide students the opportunity to learn very important concepts and techniques used by scientists in laboratories conducting biotechnology research. Some of the experimental procedures may have been modified or adapted to minimize equipment requirements and to emphasize safety in the classroom, but do not compromise the educational experience for the student. The experiments have been tested repeatedly to maximize a successful transition from the laboratory to the classroom setting. Furthermore, the experiments allow teachers

and students the flexibility to further modify and adapt procedures for laboratory extensions or alternative inquiry-based investigations.

Organizing and Implementing the Experiment

Class size, length of laboratory sessions, and availability of equipment are factors which must be considered in the planning and the implementation of this experiment with your students. These guidelines can be adapted to fit your specific set of circumstances.

If you do not find the answers to your questions in this section, a variety of re-sources are continuously being added to the EDVOTEK web site. www.edvotek.com

In addition, Technical Service is available from 9:00 am to 6:00 pm, Eastern time zone. Call for help from our knowledgeable technical staff at 1-800-EDVOTEK (1-800-338-6835).

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Please have the following information ready:

- Experiment number and title
- Kit lot number on box or tube
- Literature version number (in lower right corner)
- Approximate purchase date

Visit the EDVOTEK web site often for updated information.

Pre-Lab Preparations

A. PREPARATION OF THE DILUTED LIQUID SOAP SOLUTION

1. Add all of the of liquid soap into a beaker containing 2 L of tap water. Let it stir gently to avoid suds.
2. Dispense 140 ml of the diluted liquid soap solution into 10 plastic cups.
3. Label these cups "Control soln." Distribute one cup per group.

B. PREPARATION OF THE BICARBONATE SOLUTION

1. Transfer the entire amount of bicarbonate powder to a beaker or flask containing 2.5 L of tap water. Mix well.
2. Add 500 ml of diluted liquid soap solution (prepared above). Mix well.
3. Dispense 140 ml of the soap/bicarbonate solution into 20 plastic cups.
4. Label these cups "CO₂." Distribute two cups per group.

Each student group will receive:

- One "Control soln" cup containing the water/diluted soap solution
- Two "CO₂" cups containing the soap/bicarbonate solution
- One plastic syringe

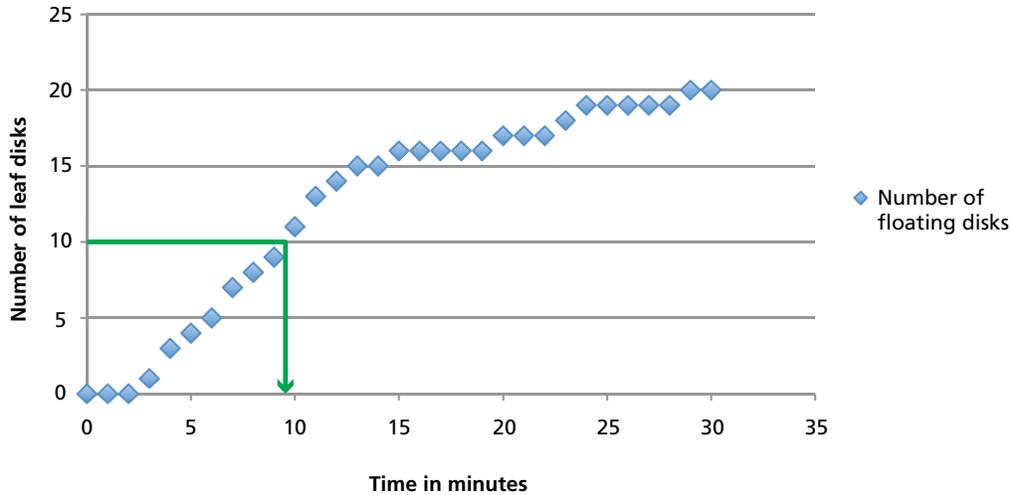
Reagents to be shared among student groups:

- Leaves (i.e. spinach, ivy, pokeweed)

Experiment Results and Analysis

Graph 1 Result - In this graph, the cup with its floating disks (spinach) was placed under the light source 15 cm away. The number of floating disks was recorded at the end of each minute until all of the disks were floating.

Graph 1 - Disks floating in 15 cm Light - CO₂ cup during 30 minutes

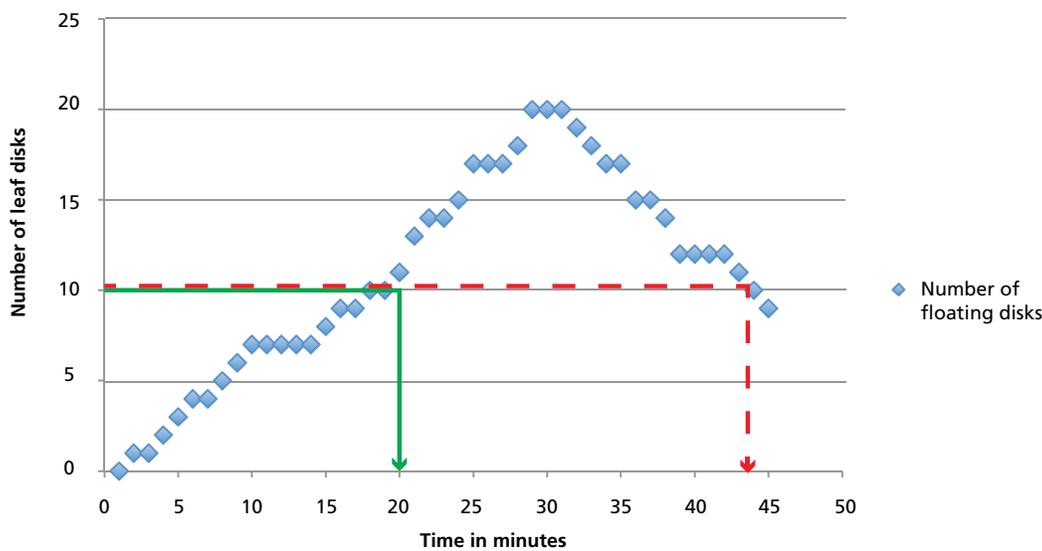


- Label the independent variable (horizontal x-axis): Time in minutes
- Label the dependent variable (vertical y-axis): Number of floating leaf disks
- Title the Graph: Disks floating in 15 cm Light - CO₂ cup during 30 minutes
By extrapolating the graph above, the ET50light value is 9.5. This means that 50% of the leaf disks (or 10 leaf disks) were floating at 9.5 minutes.

Experiment Results and Analysis

Graph 2 Result - In this graph, the light was turned off after all the leaf disks have floated for 30 minutes. The cup with its floating disks was placed in the dark. Every minute, remove the dark cover and count the number of leaf disks that were still floating in the next 15 minutes.

Graph 2 - Number of Leaf Disks



- Label the independent variable (horizontal x-axis): Time in minutes
- Label the dependent variable (vertical y-axis): Number of leaf disks
- Title the Graph: Disks floating in 30 cm Light - CO₂ cup during 30 minutes and disks sinking in 30 cm Light - CO₂ cup during 15 minutes
- By extrapolating the graph above, the ET_{50light} value is 20. This means that 50% of the leaf disks (or 10 leaf disks) were floating at 20 minutes.
- By extrapolating the graph above, the ET_{50resp} value is 44 - 30 = 14. This means that 50% of the leaf disks have sunk after the leaf disks were transferred to the dark conditions.
- Because respiration occurs in both the light and dark, the rate of photosynthesis (ET_{ps}) is the sum of the rate in the light plus the respiration rate. The rate of photosynthesis (ET_{ps}) is calculated as follows:

$$\frac{1}{ET_{50ps}} = \frac{1}{ET_{50light}} + \frac{1}{ET_{50resp}}$$

Respiration Rate: $ET_{50resp} = 44 \text{ min} - 30 \text{ minutes} = 14 \text{ minutes}$
 $1/ET_{50resp} = 1/14\text{min} = 0.07 \text{ min}^{-1}$

Rate in light: $ET_{50light} = 20 \text{ min}$
 $1/ET_{50light} = 1/20 \text{ min} = 0.05 \text{ min}^{-1}$

Photosynthesis Rate: $1/ET_{50ps} = 1/ET_{50light} + 1/ET_{50resp}$
 $1/ET_{50ps} = 0.05 \text{ min}^{-1} + 0.07 \text{ min}^{-1} = 0.12 \text{ min}^{-1}$

Photosynthesis Rate: $1.2 \times 10^{-1} \text{ min}^{-1}$

**Please refer to the kit
insert for the Answers to
Study Questions**

 Material Safety Data Sheet May be used to comply with OSHA's Hazard Communication Standard. 29 CFR 1910.1200. Standard must be consulted for specific requirements.			
IDENTITY (As Used on Label and List)		Note: Blank spaces are not permitted. If any item is not applicable, or no information is available, the space must be marked to indicate that.	
Sodium bicarbonate			
Section I			
Manufacturer's Name EDVOTEK		Emergency Telephone Number 202-370-1500	
Address (Number, Street, City, State, and ZIP Code)		Telephone Number for information 202-370-1500	
1121 5th Street NW		Date Prepared 6-12-12	
Washington DC 20001		Signature of Preparer (optional)	
Section II - Hazardous Ingredients/Identity Information			
Hazardous Components (Specific Chemical Identity; Common Name(s))	OSHA PEL	ACGIH TLV	Other Limits Recommended % (Optional)
CAS # 144-55-8	No data		
Synonyms: Sodium hydrogen carbonate; sodium acid carbonate; baking soda			
Section III - Physical/Chemical Characteristics			
Boiling Point	N.D.	Specific Gravity (H ₂ O = 1)	2.159
Vapor Pressure (mm Hg.)	N.D.	Melting Point	N.D.
Vapor Density (AIR = 1)	N.D.	Evaporation Rate (Butyl Acetate = 1)	N.D.
Solubility in Water 7.8g/100g water @ 18° C (64° F).			
Appearance and Odor Odorless, White crystalline powder			
Section IV - Fire and Explosion Hazard Data			
Flash Point (Method Used)	Flammable Limits	LEL	UEL
Extinguishing Media Use any means suitable for extinguishing surrounding fire.			
Special Fire Fighting Procedures Wear SCBA and protective clothing to prevent contact with skin and eyes			
Unusual Fire and Explosion Hazards None			
Section V - Reactivity Data			
Stability	Unstable	Conditions to Avoid	
	Stable	X	
Incompatibility (Materials to avoid) Reacts with acids to form carbon dioxide.			
Hazardous Decomposition or Byproducts Gaseous carbon dioxide.			
Hazardous Polymerization	May Occur		
	Will Not Occur	X	
Section VI - Health Hazard Data			
Route(s) of Entry:	Inhalation? Yes	Skin? Yes	Ingestion? Yes
Health Hazards (Acute and Chronic) Irritation			
Carcinogenicity: NTP? IARC Monographs? OSHA Regulation?			
Signs and Symptoms of Exposure Irritation			
Medical Conditions Generally Aggravated by Exposure			
Emergency and First Aid Procedures Inhalation: Remove to fresh air. Ingestion: Drink several glasses of water to dilute. If large swallowed, get medical advice. Skin Contact: Not expected to require first aid measures. Eye Contact: Wash thoroughly with running water. Get medical advice if irritation develops.			
Section VII - Precautions for Safe Handling and Use			
Steps to Be Taken in case Material Is Released or Spilled Ventilate area of leak or spill. Sweep up and containerize for reclamation or disposal. Vacuuming or wet sweeping may be used to avoid dust dispersal. Small amounts may be flushed to sewer with plenty of water.			
Waste Disposal Method See above			
Observe all federal, state, and local laws.			
Precautions to be Taken in Handling and Storing Keep in a tightly closed container, stored in a cool, dry, ventilated area. Protect against physical damage. Observe all warnings and precautions listed for the product.			
Other Precautions			
Section VIII - Control Measures			
Respiratory Protection (Specify Type)			
Ventilation	Local Exhaust	yes	Special
	Mechanical (General)		Other
Protective Gloves	yes	Eye Protection	Chem safety
Other Protective Clothing or Equipment			
Work/Hygienic Practices Wash thoroughly after handling			